

## Types Of Chemical Reactions Pogil Answer Key Bing

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Mg(NO<sub>3</sub>)<sub>2</sub>(aq) + 2HCl(aq) → MgCl<sub>2</sub>(aq) + CaCO<sub>3</sub>(aq) + CaCO<sub>3</sub>(aq) + 2NaCl(aq) FeS(s) + 2HCl(aq) - H<sub>2</sub>S(S) + FeCl<sub>2</sub>(aq) HCl(aq) + NaOH(aq) - H<sub>2</sub>O(l) + NaCl(aq) FeBr<sub>3</sub>(aq) + H<sub>2</sub>PO<sub>4</sub><sup>-</sup> → FePO<sub>4</sub>(s) + 3KBr(aq) The chemical equations in Model 1 contain the phase notations G, O, G, and (aq). Match.

**Chemical - LTHS Answers**  
4 POGIL™ Activities for High School Chemistry 15. A student writes the following incorrect chemical equation for a single replacement reaction be- tween lithium bromide and fluorine. 2LiBr(aq) + F 2 (g) → 2Li(s) + 2FBr(g) a. In a single replacement reaction, part of an ionic compound is removed and replaced by a

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reactions that occur daily in living things and in laboratories. Knowing the main categories of chemical reactions can help you to remember and understand them. It also helps you to recognize pattern and predict the products of many chemical reactions. Model 1: Types of Chemical Reactions Type of Reaction Symbols Chemical Equations

**POGIL "High School Drama" Model 1: Types of Chemical Reactions**  
The four sets of chemical reactions shown in Model 1 have the following general names. Discuss within your group which name belongs to which set of chemical reactions. Write the name in the appropriate place in Model 1. Single Replacement Reaction Double Replacement Reaction Synthesis Reaction Decomposition Reaction

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Chemical reactions can be classified into different categories. Four common types are synthesis, decomposition, single replacement and double replacement. Specific reactions corresponding to these general types are associated with health issues, environmental problems, and manufacturing processes. In order to use chemical reactions or evaluate their effects, you need to be able to identify the type of reaction.

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Types of Chemical Reactions POGIL Do atoms rearrange in. .... Identify the type of chemical reaction from Model 1-4 that would describe this reaction. b. Write chemical formulas under the names of the substances in the word equation. c. Balance the chemical equation. 2. Identify each of the reactions below as synthesis (S), decomposition (D ...

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17. Consider the following chemical reaction written as a word equation. diphosphorus pentoxide + water → phosphoric acid. a. Identify the type of chemical reaction from Model 1 that would describe this reaction. Synthesis. b. Write chemical formulas under the names of the substances in the word equation. P 2 O 5 + 3H 2 O → H 3 PO 4

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INFORMATION There are 5 general types of chemical reactions. Those types are synthesis (or direct combination), decomposition, single replacement, double replacement (or metathesis), and combustion. Each of the five reaction types has a unique template that helps to set it apart from the other four.

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