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## **Nelson Chemistry 12 Chapter 7**

Chemistry 12 - Chapter 7 Quiz.

True/False. Indicate whether the sentence or statement is true or false. T F. 1. Chemical equilibrium means that all chemical reactions have stopped. T F. 2. Equilibrium can only occur in a closed system.

## **Chemistry 12 - Chapter 7 Quiz - Nelson**

EQUILIBRIUM (Nelson Chemistry 12,

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Section 7.1, Pg.424) □ Some reactions occur in only one direction □ Frying an egg □  $2 \text{Pb}(\text{NO}_3)_2 \rightarrow 2 \text{PbO} + 4 \text{NO}_2 + \text{O}_2$  □ Burning a piece of paper □  $\text{C}_6\text{H}_{12}\text{O}_6(\text{s}) + 6 \text{O}_2(\text{g}) \rightarrow 6 \text{CO}_2(\text{g}) + 6 \text{H}_2\text{O}(\text{l})$  □ Other reactions can occur in both directions (reversible) □  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}(\text{s}) \leftrightarrow \text{CuSO}_4(\text{s}) + 5 \text{H}_2\text{O}(\text{l})$  □  $\text{H}_2(\text{g}) + \text{I}_2(\text{g}) \leftrightarrow 2 \text{HI}(\text{g})$  □  $\text{N}_2\text{O}_4(\text{g}) \leftrightarrow 2 \text{NO}_2(\text{g})$  EQUILIBRIUM: when the forward and reverse reactions are occurring at the same rate.

## **EQUILIBRIUM (Nelson Chemistry 12, Section 7.1 ...**

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Section 7.5: Quantitative Changes in  
Equilibrium Systems ... .. The

### **Section 7.5: Quantitative Changes in Equilibrium Systems ...**

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Chapter 7: Chemical Equilibrium 7.1-4.

Section 7.1 Questions, page 428 1. (a) A sealed bottle of pop is a dynamic equilibrium because the rate of conversion between carbon dioxide and carbonic acid is equal in both directions in the closed system. (b) When I open the pop bottle, carbon dioxide leaves the bottle, so the equilibrium conditions no longer exist and a new equilibrium must form between the pop and the atmosphere.

### **Section 7.1: Equilibrium Systems Mini Investigation: The ...**

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Electrochemistry 555 (b) Briefly describe the issue related to lead in the environment. ...

chapter Redox

Reactions 556 ...

## **Unit 7 Electrochemistry - Nelson**

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Chapter 7 – Chemical Systems and Equilibrium Section 7.1 Questions. p. 438 Question 11. Many of the nutrients in digested food, such as the carbohydrate fructose, are absorbed into the cells of the small intestine by diffusion.

## **Nelson Education - Secondary Science - Chemistry 12**

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## **Matric part 1 Chemistry, Manufacture of NaOH From Brine ...**

The values can now be substituted into the equilibrium equation for the ionization of a .  $+(-) - = + \times = -$

## **Section 8.7: Acid-Base Titration Tutorial 1 Practice, page 547**

1.1 The Fundamental Chemistry of Life  
17 5/2/12 7:20 AM 1.1 Review Summary

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- Isotopes are different forms of the same element, with different numbers of neutrons. A radioisotope is an unstable isotope that decays to release particles.
- Valence electrons are the electrons in the outermost electron shell of an atom.

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Statement: At a distance of 0.35 m, the magnitude of the electric field is  $7.2 \times 10^{-2}$  N/C and the magnitude of the electric potential is  $2.5 \times 10^{-2}$  V. 3.

## **Section 7.6: The Millikan Oil Drop Experiment**

$= 7.0 \text{ m/s [W]}$ ;  $m_2 = 60.0 \text{ g} = 0.0600 \text{ kg}$ ;  $v_{i2} = 0 \text{ m/s}$  Required:  $v_{f1}$ ;  $v_{f2}$   
Analysis: Since the second ball is initially at rest in the head-on elastic collision, use the simplified equations,  $v_{f1} = \frac{m_1 - m_2}{m_1 + m_2} v_{i1}$  and  $v_{f2} = \frac{2m_1}{m_1 + m_2} v_{i1}$ .  
Solution:  $v_{f1} = \frac{m_1 - m_2}{m_1 + m_2} v_{i1} \dots$



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## **Section 5.4: Collisions**

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Unit 1: Organic Chemistry U1-10 48. (a)  
There are 5 carbon atoms in a molecule  
of cyclopentane. (b) There are 4 carbon  
atoms ...

## **Unit 1 Review, pages 120-127**

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