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Modern Chemistry Chapter 15 Acid-Base Titration & pH Section 1 self-ionization of water occurs when two water molecules produce a hydronium (H_3O^+) and a hydroxide (OH^-) ion $2 \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$ The ionization constant (K_w) of water is: $K_w = [\text{H}_3\text{O}^+][\text{OH}^-] = 1.0 \times 10^{-14} \text{ M}$ Acidic, Basic, & Neutral IF $[\text{H}_3\text{O}^+] > [\text{OH}^-]$ then solution is acidic.

Modern Chemistry Chapter 15 Acid-Base Titration & pH

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and the Concept of pH In the space provided, write the letter of the term or phrase that best completes 15 Acid-Base Titration and pH Section: Aqueous Solutions and the Concept of pH 1 d2 3 d 4 b 5 c 6 b 7 a ...

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SECTION 3 continued 4. Table 6on page 485 of the text lists several amphoteric species, but only one other than water is neutral. NH 3 a. Identify that neutral compound. b. Write two equations that demonstrate this compound's amphoteric properties. (Answers will vary, but one equation should form NH

14 Acids and Bases

Modern Chemistry 96 Quiz Section Quiz: Aqueous Solutions and the Concept of pH In the space provided, write the letter of the term or phrase that best completes ... 15 Acid-Base Titration and pH Section: Aqueous Solutions and the Concept of pH 1. d2. 3. d 4. b 5. c 6. b 7. a 8. b 9. d10. Section: Determining pH and Titrations 1. d 2. b

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Assessment Acid-Base Titration and pH

These lecture presentations were designed for my high school Chemistry I Honors class. Students of high school and college general chemistry may find them useful as a supplement to their own class notes or as a review. Teachers, please feel free to use and modify them for your own classes.

Mrs. J's Chemistry Page - Lecture Notes

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Modern Chemistry 3 Chemical Equations & Reactions CHAPTER 8 STUDY GUIDE Chemical Equations and Reactions SECTION 8-1 SHORT ANSWER Answer the following questions in the space provided.

1. Match the symbol on the left with its appropriate description on the right.

Modern Chemistry Chapter 8 Review Chemical Equations And ...

SECTION 1 continued Date Class /o/3Qs— 20/3 5. Convert a pressure of 0.200 atm to the following units: 152 20.3 a. mm Hg b. kPa Of ROC) 6. When an explosive like TNT is detonated, a mixture of gases at high temperature is created. Suppose that gas X has a pressure of 50 atm, gas Y has a pressure of 20 atm, and gas Z has a pressure of 10 atm ...

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