

Modern Chemistry Chapter 6

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Modern Chemistry. Chapter 6. Chemical Bonding. Chemical Bond A link between atoms that results from the mutual attraction of their nuclei for electrons Electrostatic attraction between proton and electron Classified by the way the valence e- are distributed around nuclei of combined atoms Types of Bonds Ionic A chemical bond resulting from electrostatic attraction between cations and anions In a pure ionic bond, one atom completely gives up one or more e- to another atom (rarely happens) ...

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1. Modern Chemistry Chapter 6- Chemical Bonding. chemical bond- a mutual electrical attraction. between the nuclei and the valence electrons of. different atoms that binds the atoms together. 2. ionic bond- a chemical bond that results from. the electrical attraction between anions and. cations.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) positive ions. (d) negative ions.

6 Chemical Bonding

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