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Modern Chemistry Chapter 5. STUDY. PLAY. periodic law. the physical and chemical properties of the elements are periodic functions of their atomic numbers. periodic table. an arrangement of the elements in order of their atomic numbers so that elements with similar properties fall in the same column, or group. lanthanides.

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a. 2 and 4 c. 4 and 5 b. 3 and 4 d. 5 and 6 ____ 21. The discovery of the noble gases changed Mendeleev's periodic table by adding a new. a. period. c. group. b. series. d. sublevel block. ____ 22. In the modern periodic table, elements are ordered according to. a. decreasing atomic mass. c. increasing atomic number. b. Mendeleev's original ...

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Chapter 5 PRACTICE TEST. Chapter 5 PRACTICE TEST Multiple Choice Identify the letter of the choice that best completes the statement or answers the question. ____

____ 1. The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to a. Mendeleev.

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