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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode and exist either alone or in molecules containing the

Particle Mass number 19. Proton 20. Neutron 21. Electron 10 CHAPTER 3 TEST Relative charge Location MODERN CHEMISTRY HRW material copyrighted under notice appearing earlier in this work. Menu Print Name Date Class CHAPTER 3 TEST continued SHORT ANSWER Write the answers to the following questions in the space provided. 22.

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode, which is called the cathode and is emitted from the negative electrode.

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Modern Chemistry Chapter 3 1 Review Answers

3:3 mol H 2; or their reciprocals 7. Given the following equation: 4NH 3(g) 6NO(g) \rightarrow 5N 2(g) 6H 2O(g) 1 mol NO:1 mol H 2O a. What is the mole ratio of NO to NH 3? 0.360 mol c. If 0.240 mol of NH 3 react according to the above equation, how many moles of NO will be consumed? 8.

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