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force that will increase the speed of a one-kilogram mass by one meter per second each second that the force is applied. barometer. A device used to measure atmospheric pressure.

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174051167 dalton's law of partial pressures 1)states that the total pressure of a mixture of gases is equal to the sum of the pressures of all the

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gases in the mixture, 2) At constant volume and temperature, the total pressure exerted by a mixture of gases is equal to the sum of the partial pressures of the component gases.

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More than one branch may be appropriate. a. A forensic scientist uses chemistry to find information at the

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scene of a crime.

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Chapter Test B. Chapter: Stoichiometry.

PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. 1.

Knowing the mole ratio of a

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reactant and product in a chemical reaction would allow you to determine a. the energy released in the reaction.

## **Name Class Date Assessment) Chapter Test B**

Modern Chemistry 104 Chapter Test  
Chapter: Solutions In the space  
provided, write the letter of the term or

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phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. Agitation prevents settling in a(n) a. alloy. b. homogeneous mixture. c. suspension. d. gaseous mixture. \_\_\_\_\_ 2. All of the following are heterogeneous mixtures except

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