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Mechanics of Fluids, Fourth Edition Merle C. Potter, David C. Wiggert, Bassem Ramadan, and Tom I-P. Shih Publisher, Global Engineering: Christopher M. Shortt Senior Acquisitions Editor: Randall Adams Senior Developmental Editor: Hilda Gowans

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4.0 Pa ns. 1.2 The earth's atmosphere can be modeled as a uniform layer of air of thickness 20 km and average density 0.6 kg/m³ (see Table A-6). Use these values to estimate the total mass and total number of molecules of air in the entire atmosphere of the earth. Solution: Let R_e be the earth's radius ≈ 6377 km.

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Chapter 1 • Introduction 1.1 A gas at 20°C may be rarefied if it contains less than 10¹² molecules per mm³. If Avogadro's number is 6.023E23 molecules per mole, what air pressure does this represent? Solution: The mass of one molecule of air may be computed as Molecular weight 28.97 mol⁻¹ $m = \frac{28.97}{6.023 \times 10^{23}} = 4.81 \times 10^{-23}$ g Avogadro's number 6 ...

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Frank M. White is Professor of Mechanical and Ocean Engineering at the University of Rhode Island. He studied at Georgia Tech and M.I.T. In 1966 he helped found, at URI, the first department of ocean engineering in the country. Known primarily as a

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