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Chem John Erickson Answers

A copper cylinder. 12.0 cm in radius, is 44.0 cm long. If the density of copper is 8.90 g/cm³, calculate the mass in grams of the cylinder. (assume pi = 3.14) ... John Erickson. Chem Worksheet 11-1 There are a variety of units that can be used when measuring. For example, the length of an object can ...

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2SO₄ + 2Al₂(SO₄)₃ + 3H₂ - balance the equation 2 Al + 3 H₂SO₄ → 1 Al₂(SO₄)₃ + 3 H₂ ... Chem Worksheet 12-2 Stoichiometry Strategy Amount moles molar mass (g/mol) 22.4 L/mol KNOWN UNKNOWN Mass rams V olume f gas at STP iters Particles ... John Created Date:

Stoichiometry Problems Name Chem Worksheet 12-2

Chem Worksheet 4-2 Nam Any given element can have more than one isotope. To distinguish between the different isotopes of an atom, the element is named with its mass number, for example lithium-7. Remember that the mass number is the number of protons and neutrons. When symbols are used to represent an isotope

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2Cl 4 20. Carbon dioxide 9. Xenon difluoride 21. Carbon monoxide 10. S 2Cl 2 22. ClF 3 11. Iodine trichloride 23. Iodine monochloride 12. P 2S 5 24. CH 4 Naming Molecular Compounds Name _____ Chem Worksheet 9-2 Examples #1. Write the chemical formula for diphosphorus pentoxide - this compound contains two phosphorus atoms and five oxygen atoms ...

Naming Molecular Compounds Name Chem Worksheet 9-2

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2.54 cm — 45.72 cm Example Convert 25 gallons to liters. - Write the 'given' over l. - Write the units of the unknown. Insert the conversion factor so that units cancel. Solve. 25 25 gal 25 > 01 3.785 L = 95 L (rounded) Convert the following measurements using dimensional analysis. Set up problem using fractions. Cross out the units that cancel.

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