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questions 4 and 5. Check your answers with the Gizmo. (Because dissociation is not always complete, your answers may vary slightly from values in the Gizmo.) [H 2 SO 4] = 0.108 M pH H 2 SO 4 = 0.967 [HBr] = 0.210 M pH HBr = 0.678 7. Apply: For a strong base, the concentration of hydroxide ions [OH-] is roughly estimated to

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neutralize, pH, strong acid, strong base, titrant, titration, titration curve, weak acid, weak base Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

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the Gizmo. (If there is no coefficient, the value is 1....so type in 1's for these boxes for this example :) • Record the appropriate volumes in the “mL NaOH” and “mL HBr” boxes. In other words, what was your answer to #9 and how much liquid is in the flask? • Record the concentration of the titrant in the M NaOH box.

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