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The only explanation for the observed faster cooling is that the refrigerator might be a non-self defrost model which accumulated an ice layer on the freezing coils. Then, when the hot water tray was placed on the ice layer, it melted and reduced the thermal insulation between the cooling coil and the ice tray.

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Most heat transfer is by convection.  $1-30 Q = \dot{m} c_p (T_2 - T_1)$   $Q_{conv} = hA(T_s - T_\infty)$  From Table 1-1  $h = 4-5 \text{ W/m}^2 \cdot \text{K}$   $Q_{rad} = \epsilon \sigma A (T_s^4 - T_\infty^4)$   $\epsilon = 0.4$   $\sigma = 5.669 \times 10^{-8} \text{ W/m}^2 \cdot \text{K}^4$   $Q_{total} = Q_{conv} + Q_{rad} = 24.3 \text{ W} + 28.7 \text{ W} = 53 \text{ W}$  Convection and radiation are about the same magnitude.  $Q_{total} = \dot{m} c_p (T_2 - T_1) = 0$  (insulated)  $Q_{conv} = hA(T_s - T_\infty)$

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The minus sign indicates, heat flow will be in the direction of decreasing temperature. Ignore the minus sign and re-arrange the equation to obtain the relation for temperature difference. Step 2 of 2 Substitute 3 kW for Q, for A, 2.5 cm for and for k in equation (1) to obtain the value of temperature difference.

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