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are equal.

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period Increases down  
a family Atomic Mass  
Increases across a  
period Increases down  
a family Atomic Radius  
 $\frac{1}{2}$  the distance  
between nuclei of two  
like atoms . Decreases  
across a period (due to  
increased attraction of  
more positive nuclei to  
electron clouds)  
increases down a  
family (due to addition  
of an energy level)  
Electron Affinity  
amount of energy



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released when an atom  
gains an electron  
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in the Periodic Table  
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26m 9s Trends on the  
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Students learn to describe the pattern in atomic number, atomic mass, atomic radius, ionization energy, and electron affinity as they look across a period and down a family of the periodic table.

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from object ...

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