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Felder, Rousseau: Elementary Principles of Chemical ...

Solution From the first law of thermodynamics applied for a closed system, the heat transferred is given by: $Q = \Delta U$ The change in internal energy is defined by Equation 8.3-6 of Felder & Rousseau: $\Delta U = \int_{T_1}^{T_2} C_v(T) dT$ Substituting the heat capacity equation into the integral:

Chapter 8 Balances on Nonreactive Processes Date:

Richard M. Felder, Ronald W. Rousseau, Richard M Felder
Elementary Principles of Chemical Processes, 2005 Edition
Integrated Media and Study Tools, with Student Workbook 3rd
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CHAPTER TWO - College of Engineering and Applied Science

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