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Ivanpenaloza. chem study guide chapter
6 (12/5/12) Terms in this set (53) the
idea of arranging the elements in a table

according to their chemical and physical properties is attributed to \_\_\_\_\_.

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Chemistry Chapter 6 Study Guide. STUDY. PLAY. Periodic Law. states that the repeating chemical and physical properties of elements change

periodically with the atomic numbers of the elements. Newland's Law of Octaves-Was never excepted by fellow scientists-Organized Elements by increasing atomic mass.

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Mendeleev 6. atomic mass 7. elements 8. atomic number 9. Henry Moseley 10. protons 11. periodic law

#### **Ch 6 Study Guide answers**

Unit 2 CHAPTER 6 Chemistry in Biology . 15 Name Date Class In your textbook, read about the building blocks of life. For each statement below, write true or

false. true 1. Carbon atoms can bond together in straight chains, branched chains, or rings. false 2. Large molecules containing carbon atoms are called micromolecules. false 3.

#### Name

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letter of the choice that best completes the statement or answers the question.

1. Atomic radii cannot be measured directly because the electron cloud surrounding the nucleus does not have a clearly defined

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must conserve the balance of mass. identify which number indicates the number of atoms of each element in a molecule. the number 6. minimum amount of energy required for reactants to form products. activation energy. substance that lowers energy needed to start a chemical reaction. catalyst.

#### study guide chapter 6 section 2 chemical reactions ...

basic chemistry, the structure and function of cells, and cell ener-getics. In Chapter 6, students learn the basic concepts of chemistry that are important in biology. Chapter 7 introduces cell structure and function of organelles. This discussion is expanded upon in Chapter

8 through an in-depth view of cel-lular transport and the cell cycle.

Chapter 6: The Chemistry of Life Chemistry Chapter 6 Study Guide. arranged elements according to atomic mass and used the arrangement to predict the properties of missing elements. elements that are

characterize by the filling of p orbitals are classified as \_\_\_\_. which subatomic particle plays the greatest part in determining the properties of an element.

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Chapter 6 Thermochemistry Study

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Chemistry Section Goals and Introductions As always, the Review Skills section for this chapter is important. A quick review of atomic mass and moles in Section 3.6, the information on ionic and molecular compounds in Section

#### **Chapter 6 More on Chemical**

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#### **Compounds**

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compounds. Be able to identify organic molecules? answerOrganic compounds are made primarily of carbon. Each small

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