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## **Chapter 15 Water And Aqueous**

Chapter 15 Water and Aqueous Systems 159 SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449) This section describes the properties of water in the liquid and solid states and explains how hydrogen bonding affects the surface tension and vapor pressure of water. Water in the Liquid State (pages 445–447) 1.

## **SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449)**

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Chapter 15 Water and Aqueous Systems. Chapter 15 "Water and Aqueous Systems". The Water Molecule: a Review. • Water is a simple tri-atomic molecule, H<sub>2</sub>O. • Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values) • bond angle of water = 105°.

## **Chapter 15 Water and Aqueous Systems**

Chemistry, Chapter 15, Water and Aqueous Systems. surface tension. surfactant. aqueous solution. solvent. the inward force or pull that tends to minimize the surface ar... any substance that interferes with hydrogen bonding between wa.... is water

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that contains dissolved substances.

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Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous Aqueous Systems - 15.3 Lesson Check - Page 507: 21 Answer

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Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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Review. Surface tension is the \_\_\_\_\_. How does the surface tension of water compare with the surface tensions of most other liquids? Which type of mixture(s) exhibit the ...

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## **Chapter 15 water and aqueous systems - LinkedIn SlideShare**

Chapter 15 "Water and Aqueous Systems" Tools. Copy this to my account; E-mail to a friend; Find other activities ... a relatively strong intermolecular force responsible for water's high surface tension: suspension: a mixture from which particles settle out upon standing: aqueous solution: a solution in which the solvent is water: solvent: the ...

## **Quia - Chapter 15 "Water and Aqueous Systems"**

Chapter 15 - Water and Aqueous Systems - 15.1 Water and Its Properties - 15.1 Lesson Check - Page 493: 2 Answer The structure of ice is a regular open framework of water molecules arranged like a honeycomb.

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## **Chemistry (12th Edition) Chapter 15 - Water and Aqueous**

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SECTION 15.1 WATER AND ITS PROPERTIES (pages 445–449)

CHAPTER 15, Water and Aqueous Systems (continued) 6 Circle the letter next to each sentence that describes a result of the surface tension of water a In a full glass of water, the water surface seems to bulge over the rim of the glass b Water beads up into

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