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15 Acids Bases

Test Modern
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Chapter 15 Acids Bases Test

An ionic compound

composed of a cation

(+) from a base and an

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anion (-) from an acid.
When a strong acid
and a strong base
react they produce a
Neutral salt and water

Chemistry Chapter 14, 15 Acids and Bases Test Review

...

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Period Honors

Chemistry — Chapter
15 Acid & Base Test.

Matching: L1. strong
acids A. bitter L2. pH B.

ionize only slightly in
solution £3. titration C.

salt + H₂ (g) F 4.

neutralization D. salt +
water. 5. bright pink

determining the
concentration of an

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acid with a known concentration of a base. 5 6. acid + metal
F. reaction of an acid with a base hydronium ion
G. color of bases in phenolphthalein

Ch15 Test_key.pdf - Google Docs

1. What is NOT true about buffered solutions that is true about other solutions in equilibrium? The $[H^+]$ does not equal... 2. A 60mL HCl solution is

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titrated with 25mL of a
0.60M KOH solution.

What is the
concentration of the
HCl solution? 2.5... 3.

What are the
hydronium and
hydroxide ion ...

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Chapter 15: Acids and Bases

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Define acid and base according to Arrhenius's theory of ionization. Explain the differences between strong and weak acids and bases. FIGURE 15-1 (a) Fruits and fruit juices contain acids such as citric acid and ascorbic acid.

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Carbonated beverages contain benzoic acid, phosphoric acid, and carbonic acid. (b) Many household cleaners contain ...

CHAPTER 15 Acids and Bases

when acids or bases act as acids or bases, they form their conjugate and the strength is the inverse of their parent. and there are certain strong acids and strong

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bases. LiOH , NaOH ,
 KOH - when act as
bases (get rid of H^+
ion) then left with Li
ion, Na ion, K ion.

Chem 2 (Quiz 2): Chapter 15 (Acids and Bases) Flashcards ...

a. a strong acid b. a
weak acid c. a strong
base d. a weak base e.
neither an acid nor a
base 24) A 25.0-mL
sample of a solution of
a monoprotic acid is

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titrated with a 0.115 M NaOH solution. The titration curve above was obtained. The concentration of the monoprotic acid is about _____ mol/L.

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Chemistry

**acids bases salts
chapter 15**

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Chapter 15 Aqueous
Equilibria: Acids and
Bases. Instructor's
Resource Materials
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Chemistry, ...

Conjugate Acid-Base
Pairs: Chemical species
whose formulas differ
only by one hydrogen
ion, H^+ Brønsted-

Lowry Acid: A

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Chemistry
substance that can transfer hydrogen ions, H^+ . In other words, a proton donor

Chapter 15 Equilibria: Acids and Bases

Chapter 15: Acids and Bases
Acids and Bases
Arrhenius Definitions:
acids - compounds that produce an increase in $[H^+]$ when dissolved in water
bases - compounds that produce an increase in

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[OH⁻] when dissolved
in water Lewis

Definitions: acids -
electron pair acceptors
bases - electron pair
donors Brønsted-Lowry
Definitions: acids - H⁺
donors

Chapter 15: Acids and Bases Acids and Bases

Modern Chemistry 127
Chapter Test Name
Class Date Chapter
Test A, continued

_____ 14. A strong base

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in an aqueous solution
a. is a weak
electrolyte. b. produces
many H ions. c. will not
dissolve. d. completely
dissociates into ions.

____ 15. In a Brønsted-
Lowry acid-base
reaction, what are
transferred from one
reactant to another? a.
electrons ...

Assessment Chapter Test A - Ed W. Clark High School

SAMPLE TEST 2
Page 16/25

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CHAPTER 15 Acids &
Bases CHAPTER 15 1.
Acids and bases can be
defined in several
ways. Which of the
following are
definitions of bases
according to these
definitions? a.a
compound the
produces hydronium
ions in water b.a
proton acceptor c. an
electron pair acceptor
d.an electron pair
donor e.both b and d
statements. 2.

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**ch 15 acids and
bases - SAMPLE
TEST 2 CHAPTER 15
Acids ...**

Chemistry: A Molecular
Approach, 3e (Tro).

Chapter 15 Acids and
Bases Multiple Choice
Questions 1) _____ is
found in carbonated
beverages due to the
reaction of carbon
dioxide with water.

**Chapter 15 Acids
and Bases - eBooks,**
Page 18/25

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Test Modern Acid Chemistry **Academic Notes and More**

Acids and Bases:

Chapter 14 & 15 . HW:

- Read Ch 14:
- Fill in as much of the acid base table as you can, as you read . Acid base conductivity and reactivity Conductivity, Reac'vity,, ... 3. Work on ch 14-15 review (due Wednesday, test Friday!) • Solutions in which [H

Acids and Bases:

Read PDF Chapter 15 Acids Bases

Chapter 14 & 15

a. Acid Base Practice

Test Answer Section

MULTIPLE CHOICE 1.

ANS: B DIF: I OBJ:

15-1.1 2. ANS: C DIF: I

OBJ: 15-1.1 3. ANS: A

DIF: I OBJ: 15-1.1

Acid Base Practice Test - Houston Independent School

...

A.P. Chemistry Practice

Test: Ch. 14, Acids and

Bases Name _____

MULTIPLE CHOICE.

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Test Modern
Chemistry

Choose the one alternative that best completes the statement or answers the question. ... None of the above will act as a weak base in water.

26) Determine the pH of a 0.15-M aqueous solution of KF. For hydrofluoric acid, $K_a = 7.0 \times 10^{-4}$. A)5.83 B)6.59 C)8.16 D)2 ...

**A.P. Chemistry
Practice Test: Ch.
14, Acids and Bases**

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Acids & Bases Chapter
Exam Instructions.

Choose your answers
to the questions and
click 'Next' to see the
next set of questions.
You can skip questions
if you would like and
come back to them ...

Acids & Bases - Practice Test Questions & Chapter Exam ...

AP Chemistry Study
Guide: Chapter 14:
Acids and Bases. and

Read PDF Chapter 15 Acids Bases

Chapter 15: Aqueous
and Acid-Base

Equilibria. Students
should be able to...

Know properties of
acids and bases . Know
definitions of acids and
bases according to
Arrhenius, and
Bronsted-Lowry.

Identify Bronsted-
Lowry conjugate
acid/base pairs.

Identify polyprotic
acids and bases.

AP Chemistry Study

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Test Modern Chemistry **Guide: Chapter 14, Acids and Bases and**

...

Modern Chemistry 131

Chapter Test Name

Class Date Chapter

Test B, continued Write

the formula for each of

the following acids in

the space provided. 22.

hydrosulfuric acid 23.

nitric acid 24.

phosphorous acid 25.

perchloric acid Refer to

the equation below to

answer questions 26

and 27. $\text{HCl(g)} + \text{NH}_3$

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26. List the y.
 $3(l) \leftrightarrow NH_4^+(aq) Cl^-(aq)$

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ecf8427e.