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17 Terms. nava342380. Chapter 12 - Mastering Chemistry Homework. Identify the type of interactions invol.... Identify the type of interactions invol.... Identify the type of interactions invol.... The separation of solute particles (ΔH1.... Solvent-solvent interactions.... -interactions between the water m....

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Chapter 12 Study Guide. The calculation of quantities in chemical reactions. The quantities to consider in a chemical reaction? On the left side of a chemical equation, you have the? On the right side of a chemical equation, you have the? In every chemical reaction, these two things are conserved. In a majority of chemical reactions, these three things are NOT conserved.

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Chapter 12 Molecular Structure  Review Skills 12.1 A New Look at Molecules and the Formation of Covalent Bonds  The Strengths and Weaknesses of Models  The Valence-Bond Model 12.2 Drawing Lewis Structures  General Procedure  More Than One Possible Structure 12.3 Resonance

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25 terms. jakydv. chapter 12 study guide chemistry. law of conservation of mass. formula mass. law of definite proportions. limiting reactant. no detectable change in mass occurs during a chemical reaction. the sum of the atomic masses given in the formula of a substan....

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Section 12.1 Gases 1. motion 2. a. small b. forces c. random d. elastic; kinetic 3. KE 1/2 mv<sup>2</sup> 4. Temperature 5. true 6. true 7. false 8. true 9. true 10. false 11. true 12. false 13. a 14. a 15. d 16. d 17. b 18. b 19. b 20. barometer 21. Evangelista Torricelli 22. gravity and atmospheric pressure 23. The atmospheric pressure has increased. Section 12 .2 Forces of Attraction 1. intermolecular forces 2. intermolecular forces

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