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Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons. ©Instructional Fair, Inc. 3. 5. 6. o + o (O2) c + o (CO2) H + O (H2O) Chemistry if8766. TYPES OF CHEMICAL BONDS Name Classify the following compounds as ionic (metal + nonmetal), covalent (nonmetal + nonmetal) or both (compound containing a polyatomic ion).

Chemistry 301

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Chemistry If8766 . IONIC BONDING Name Ionic bonding occurs when a metal transfers one or more electrons to a nonmetal in an effort to attain a stable octet of electrons. For example, the transfer of an electron from sodium to chlorine can be depicted by a Lewis dot diagram.

COVALENT BONDING Name Covalent bonding occurs when two or ...

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