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11—2 Review and Reinforcement Solving Stoichiometry Problems If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line provided. 1. 2. 5. The major types of stoichiometry problems are mass-mass, mass-volume, and volume-volume.

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II—I Review and Reinforcement (continued) Solve each of the following problems as directed. Show all your work. 13. How many moles Of magnesium are required to react with 2.0 mol of hydrochloric acid? The equation for this reaction is Mg + 2HCI -+ MgC12 + H '2 Omo\ 14. Aluminum reacts with HCl to produce aluminum chloride (AIC13) and hvdrogen …

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19. If 6.57 g of iron react with an excess of hydrochloric acid, HCl, then 11.2 g of iron(II) chloride are obtained in addition to hydrogen gas. Find the theoretical and percent yields. $Fe + 2 HCl \rightarrow FeCl 2 + H 2$ 20. If 5.45 g of potassium chlorate are decomposed to form potassium chloride, 1.75 g of oxygen gas are also given off.

Stoichiometry Review **YOU MUST SHOW ALL WORK AND YOUR ...
3 1 Page 11 Chapter 3 3 1 Review And Reinforcement … 11—1 Review and Reinforcement Stoichiometry Complete the following sentences by 'filling in the appropriate word or phrase from the list below. Each word or phrase maybe used once, more than once, or not at all reactants molar ratio quantitative actual particles. consdrvption of matter

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